

RAWATAN LARUT RESAPAN MENGGUNAKAN PENJERAP MEDIA
KOMPOSIT KARBON TERAKTIF, ZEOLIT, BATU KAPUR DAN
ENAPCEMAR SISA KERTAS

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Tesis ini dikemukakan sebagai
memenuhi syarat penganugerahan
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Universiti Tun Hussein Onn Malaysia

JUN 2019

Untuk papa dan mama tersayang



PENGHARGAAN

Alhamdulillah, syukur ke hadrat Allah S.W.T kerana dengan limpah kurnia dan izin-Nya dapatlah saya menyiapkan tesis ini bagi memenuhi keperluan Ijazah Doktor Falsafah, Universiti Tun Hussein Onn Malaysia.

Setinggi-tinggi penghargaan dan terima kasih diucapkan kepada Prof. Madya Dr. Zawawi bin Daud selaku penyelia utama, serta penyelia bersama yang lain iaitu Prof. Hj. Ab Aziz Bin Abdul Latiff dan Prof. Madya Dr. Azhar Bin Abdul Halim, atas bimbingan dan dorongan yang diberi sepanjang tempoh penyelidikan tesis ini. Penghargaan dan terima kasih juga saya ucapkan kepada rakan-rakan seperjuangan dan ahli keluarga yang banyak memberi sokongan dan bantuan bagi melengkapkan penyelidikan ini.

Akhir sekali, terima kasih diucapkan kepada semua pihak yang telah membantu saya sama ada secara langsung atau tidak langsung sepanjang tempoh penyelidikan ini. Semoga Allah S.W.T memberikan balasan yang lebih baik jua. Amin.

ABSTRAK

Kajian ini dijalankan bagi meneroka potensi WPS dan LS untuk menggantikan sebahagian AC dan ZEO dalam penghasilan media komposit baharu. Nisbah optimum AC-WPS (media hidrofobik) adalah 2:2, manakala ZEO-LS (media hidrofilik) adalah 25:15. Nisbah optimum 4:4 media hidrofobik-hidrofilik dipilih berdasarkan ciri-ciri penjerapan $\text{NH}_3\text{-N}$ dan COD, dan 30 % OPC telah digunakan sebagai bahan pengikat. Media komposit ZELPA telah dihasilkan dan pencirian sifat fizikal-kimia dilakukan. ZELPA digunakan dalam kajian isoterma penjerapan $\text{NH}_3\text{-N}$, COD dan warna pada keadaan optimum pH 7, kelajuan goncangan 200 rpm, saiz partikel 1.18-2.36 mm dan masa sentuhan 120 minit. Setelah ujian kelompok dilaksanakan, ujikaji isoterma penjerapan dan kinetik penjerapan dilakukan. Kapasiti penjerapan untuk kesemua parameter (dalam mg/g) masing-masing 27.55, 48.08 dan 34.48. Kajian perbandingan menunjukkan keupayaan penjerapan ZELPA terhadap $\text{NH}_3\text{-N}$ dan COD adalah lebih baik daripada ZEO dan AC, dan setanding dengan AC bagi penjerapan warna. Kajian kinetik penjerapan mendapati ZELPA mengikuti hampir kesemua model yang dikaji iaitu pseudo tertib pertama dan kedua, Elovich dan pembauran intra-partikel. Kajian penjerapan terus dengan menggunakan model EBCT menunjukkan penyingkiran sebanyak 99 % untuk $\text{NH}_3\text{-N}$ dan 98 % untuk COD dan warna. Kapasiti bolos dan masa tepu didapati makin berkurangan dengan berlakunya peningkatan terhadap kadar aliran. Didapati data uji kaji lebih mematuhi Model Thomas dan Yoon-Nelson berbanding Adams-Bohart. ZELPA boleh digunakan semula selepas menjalani ujikaji regenerasi dengan larutan 0.5 M NaCl pada pH 11-12 (dilaras dengan 1 M NaOH). Dengan kapasiti penjerapan untuk $\text{NH}_3\text{-N}$, COD dan warna (dalam mg/L) masing-masing 32.26, 38.76 dan 31.95. Gabungan media alternatif WPS dan LS menunjukkan impak positif dalam penjerapan larut resapan. Penggunaan WPS dalam teknologi penjerapan secara tidak langsung berupaya menangani masalah pelupusan sisa kertas yang tidak boleh dikitar semula.

ABSTRACT

This study was conducted to explore the potential of WPS and LS to partially replace AC and ZEO in producing a new composite adsorbent. Optimum ratio of AC-WPS (hydrophobic media) was 2:2, while ZEO-LS (hydrophilic media) was 25:15. Optimum ratio 4:4 for hydrophobic-hydrophilic media were chosen according to adsorption behaviour of $\text{NH}_3\text{-N}$ and COD, and 30 % of OPC was applied as a binder. Composite media ZELPA was produced and their physico-chemical characteristic were determined. ZELPA was used in isotherm study for $\text{NH}_3\text{-N}$, COD and colour at optimum pH 7, 200 rpm shaking speed, 1.18-2.36 mm particle size and 120 minutes contact time. After conducting batch experiment, the adsorption isotherm and kinetic experiment were performed, as comparison between those two experiments. The adsorption capacity (in mg/g) of aforementioned parameters are 27.55, 48.08 and 34.48, respectively. Comparative study shows the adsorption capacity of ZELPA towards $\text{NH}_3\text{-N}$ and COD were improved than ZEO and AC, and on a par with AC for colour adsorption. Adsorption kinetic study found ZELPA was followed almost all kinetics models (i.e. pseudo-first and second order, Elovich and intra-particle diffusion), where pseudo-second order shows the most dominant for all parameters. Column adsorption studies using EBCT model indicates the removal for $\text{NH}_3\text{-N}$ was 99 %, while for COD and color were 98 %. The breakthrough capacity and saturated time were found to decreased with increased of flow rate. Experimental data were found more compliant with Thomas and Yoon-Nelson models, compared to Adams-Bohart model. ZELPA can be reused after undertaking regeneration process with 0.5M NaCl solution at pH 11 – 12 (adjusted by 1M NaOH). Its adsorption capacities (in mg/g) for $\text{NH}_3\text{-N}$, COD and colour were 32.26, 38.76, 31.95, respectively. Combination of WPS and LS shows a positive impact in leachate adsorption technology. WPS use in adsorption technology indirectly solve the paper waste disposal that can't be recycled.

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RUJUKAN

- Abas, S. N. A., Ismail, M. H. S., Kamal, M. L. & Izhar, S. (2013). Adsorption process of heavy metals by low-cost adsorbent: A review. *World Applied Sciences Journal*, 28(11), pp. 1518 - 1530.
- Abbas, A. A., Guo, J., Liu, Z. P., Pan, Y. Y. & Wisaam, S. A. (2009). Review on landfill leachate treatments. *Journal of Applied Sciences Research*, 5(5), pp. 534 - 545.
- Abdolali, A., Guo, W. S., Ngo, H. H., Chen, S. S., Nguyen, N. C. & Tung, K. L. (2014). Typical lignocellulosic wastes and by-products for biosorption process in water and wastewater treatment: A critical review. *Bioresource Technology*, 160, pp. 57 - 66.
- Abdullah, R., Ishak, C. F., Kadir, W. R. & Abu Bakar, R. (2015). Characterization and feasibility assessment of recycled paper mill sludges for land application in relation to the environment. *International Journal of Environmental Research and Public Health*, 12(8), pp. 9314 - 9329.
- Adeleke, A. O., Latiff, A. A. A., Gheethi, A. A. A. & Daud, Z. (2017). Optimization of operating parameters of novel composite adsorbent for organic pollutants removal from POME using response surface methodology. *Chemosphere*, 174(2017), pp. 232 - 242.
- Adhikari, B. & Khanal, S. N. (2015). Qualitative study of landfill leachate from different ages of landfill sites of various countries including Nepal. *IOSR Journal of Environmental Science, Toxicology and Food Technology (IOSR-JESTFT)*, 9(1), pp. 23 - 36.

- Agyei, N. M. (2004). *The removal of phosphate ions from aqueous solution by fly ash, slag, ordinary portland cement and related blends*. University of Pretoria: Ph.D. Thesis.
- Ahmad, A. A. & Hameed, B. H. (2010). Fixed-bed adsorption of reactive azo dye onto granular activated carbon prepared from waste. *Journal of Hazardous Materials*, 175(1-3), pp. 298 - 303.
- Ahmadi, B. & Al-Khaja, W. (2001). Utilization of paper waste sludge in the building construction industry. *Resources, Conservation and Recycling*, 32(2), pp. 105 - 113.
- Ahmadi, S. & Mostafapour, F. K. (2017). Tea waste as a low cost adsorbent for the removal of COD from landfill leachate: Kinetic Study. *Journal of Scientific and Engineering Research*, 4(6), pp. 103 - 108.
- Ahmedna, M., Marshall, W. E. & Rao, R. M. (2000). Surface properties of granular activated carbons from agricultural by-products and their effects on raw sugar decolorization. *Bioresource Technology*, 71(2), pp. 103 - 112.
- Aksu, Z. & Gönen, F. (2004). Biosorption of phenol by immobilized activated sludge in a continuous packed bed: prediction of breakthrough curves. *Process Biochemistry*, 39(5), pp. 599 - 613.
- Alam, M. Z., Ameem, E. S., Muyibi, S. A. & Kabbashi, N. A. (2009). The factors affecting the performance of activated carbon prepared from oil palm empty fruit bunches for adsorption of phenol. *Chemical Engineering Journal*, 155(1-2), pp. 191 - 198.
- Alias, N. H. M., Halim, A. A. & Wahab, M. I. A. (2011). Penyingkiran boron daripada larutan akuas menggunakan penjerap komposit berasaskan karbonmineral. *Sains Malaysiana*, 40(11), pp. 1271 - 1276.
- Alvarez-Vazquez, H., Jefferson, B. & Judd, J. S. (2004). Membrane bioreactors vs. conventional biological treatment of landfill leachate: A brief review, *Journal of Chemical Technology and Biotechnology*, 79(10), pp. 1043 - 1049.

- American Public Health Association (APHA), American Water Works Association (AWWA), Water Environment Federation (WEF) (2012). Standard Methods for the Examination of Water and Wastewater. Edisi ke-21. Washington, DC.
- Andrejkovičová, S., Sudagar, A., Rocha, J., Patinha, C., Hajjaji, W., Silva, E. F. D., Velosa, A. & Rocha, F. (2016). The effect of natural zeolite on microstructure, mechanical and heavy metals adsorption properties of metakaolin based geopolymers. *Applied Clay Science*, 126, pp. 141 - 152.
- Anisuzzaman, S. M., Joseph, C. G., Yap, Y. H. T., Krishnaiah, D. & Tay, V. V. (2015). Modification of commercial activated carbon for the removal of 2,4-dichlorophenol from simulated wastewater. *Journal of King Saud University – Science*, 27(4), pp. 318 - 330.
- Areeprasert, C., Zhao, P., Ma, D., Shen, Y. & Yoshikawa, K. (2014). Alternative solid fuel production from paper sludge employing hydrothermal treatment. *Energy Fuels*, 28(2), pp. 1198 - 1206.
- ASTM D4607-14, Standard Test Method for Determination of Iodine Number of Activated Carbon, ASTM International, West Conshohocken, PA.
- ASTM D2330-02, Standard Test Method for Methylene Blue Active Substances (Withdrawn 2011), ASTM International, West Conshohocken, PA.
- Ataman, E., Andersson, M. P., Ceccato, M., Bovet, N. & Stipp, S. L. S. (2016). Functional group adsorption on calcite: ii. nitrogen and sulfur containing organic molecules. *The Journal of Physical Chemistry C*, 120(30), pp. 16597 - 16607.
- Aziz, H. A., Adlan, M. N., Zahari, M. S. M. & Alias, S. (2004). Removal of ammoniacal nitrogen (N-NH₃) from municipal solid waste leachate by using activated carbon and limestone. *Waste Management & Research*, 22(5), pp. 371 - 375.
- Aziz, A. H., Alias, S., Adlan, M. N., Asaari, F. A. H. & Zahari, M. S. (2007). Colour removal from landfill leachate by coagulation and flocculation processes. *Bioresource Technology*, 98(1), pp. 218 - 220.

- Aziz, H. A., Foul, A. A., Isa, M. H. & Hung, Y. T. (2010). Physico-chemical treatment of anaerobic landfill leachate using activated carbon and zeolite-batch and column studies. *International Journal of Environment and Waste Management*, 5(3-4), pp. 269 - 285.
- Aziz, H. A., Hin, L. T., Adlan, M. N., Zahari, M. S., Alias, S., Ahmed, Foul, A. A. M. A., Selamat, M. R., Bashir, M. J. K., Yusoff, M. S. & Umar, M. (2011). Removal of high-strength colour from semi-aerobic stabilized landfill leachate via adsorption on limestone and activated carbon mixture. *Research Journal of Chemical Sciences*, 1(6), pp. 1 - 7.
- Aziz, H. A., Othman, N., Yusuff, M. S., Basri, D. R. H., Asaari, F. A. H., Adlan, M. N., Othman, F., Johari, M. & Perwira, M. (2001). Removal of copper from water using limestone filtration technique-determination of mechanism of removal. *Environment International*, 26(5-6), pp. 395 - 399.
- Aziz, H. A., Yusoff, M. N., Adlan, M. N., Adnan, N. H. & Alias, S. (2004). Physico-chemical removal of iron from semi-aerobic landfill leachate by limestone filter. *Waste Management*, 24(4), pp. 353 - 358.
- Aziz, S. Q., Aziz, H. A., Yusoff, M. S., Bashir, M. J. K. & Umar, M. (2010). Leachate characterization in semi-aerobic and anaerobic sanitary landfills: A comparative study. *Journal of Environmental Management*, 91(12), pp. 2608 - 2614.
- Azmi, N., Bashir, M. J. K., Sethupathi, S. & Ng, C. A. (2016). Anaerobic stabilized landfill leachate treatment using chemically activated sugarcane bagasse activated carbon: kinetic and equilibrium study. *Desalination and Water Treatment*, 57(9), pp. 3916 - 3927.
- Azmi, N., Bashir, M. J. K., Sethupathi, S., Wei, L. J. & Aun, N. C. (2015). Stabilized landfill leachate treatment by sugarcane bagasse derived activated carbon for removal of color, COD and $\text{NH}_3\text{-N}$ - Optimization of preparation conditions by RSM. *Journal of Environmental Chemical Engineering*, 3(2), pp. 1287 - 1294.

- Azouaou, N., Sadaoui, Z. & Mokaddem, H. (2014). Removal of lead from aqueous solution onto untreated coffee grounds: A fixed-bed column study. *Chemical Engineering Transactions*, 38, pp. 151 - 156.
- Babic, B., Kokunešoski, M., Gulicovski, J., Prekajski, M., Pantić, J., Mihajlović, A. R. & Matović, B. (2011). Synthesis and characterization of carbon cryogel/zeolite composites. *Processing and Application of Ceramics*, 5(2), pp. 91 - 96.
- Badmus, M. A. O. & Audu, T. O. K. (2009). Periwinkle shell: Based granular activated carbon for treatment of chemical oxygen demand (COD) in industrial wastewater. *The Canadian Journal of Chemical Engineering*, 87(1), pp. 69 - 77.
- Baiju, A., Gandhimathi, R., Ramesh, S. T. & Nidheesh, P. V. (2018). Combined heterogeneous Electro-Fenton and biological process for the treatment of stabilized landfill leachate. *Journal of Environmental Management*, 210, pp. 328 - 337.
- Bandura, L., Franus, M., Józefaciuk, G., & Franus, W. (2015). Synthetic zeolites from fly ash as effective mineral sorbents for land-based petroleum spills cleanup. *Fuel*, 147, pp. 100 - 107.
- Baral, S. S., Das, N., Ramulu, T. S., Sahoo, S. K., Das, S. N. & Roy Chaudhury, G. (2009). Removal of Cr(VI) by thermally activated weed *Salvinia cucullata* in a fixed-bed column. *Journal of Hazardous Materials*, 161(2-3), pp. 1427 - 1435.
- Barlaz, M. A. & Ham, R. K. (1993). Leachate and gas generation: In *Geotechnical Practice for Waste Disposal*, D.E. Daniel, ed., Chapman and Hall, London, pp. 113 - 136.
- Bashir, M. J. K., Aziz, H. A., Yusoff, M. S. & Adlan, M. N. (2010). Application of response surface methodology (RSM) for optimization of ammoniacal nitrogen removal from semi-aerobic landfill leachate using ion exchange resin. *Desalination*, 254(1/3), pp. 154 - 161.
- Bashir, M. J. K., Aziz, H. A., Yusoff, M. S. & Aziz, S. Q. (2012). Color and chemical oxygen demand removal from mature semi-aerobic landfill leachate using anion-

exchange resin: An equilibrium and kinetic study. *Environmental Engineering Science*, 29(5), pp. 297 - 305.

Baykal, B. B. & Guven, D. A. (1997). Performance of clinoptilolite alone and in combination with sand filters for the removal of ammonia peaks from domestic wastewater. *Water Science and Technology*, 35(7), pp. 47 - 54.

Bentoulfa, S., Fayala, F., Cheikhrouhou, M. & Ben Nasrallah, S. (2007). Porosity determination of jersey structure. *AUTEX Research Journal*, 7(1), pp. 63 - 69.

Berge, N. D., Reinhart, D. R. & Townsend, T. G. (2005). The fate of nitrogen in bioreactor landfills. *Critical Reviews in Environmental Science and Technology*, 35(4), pp. 365 - 399.

Bhat, I. U. H., Mungkar, A. N., Lee, K. E. & Khanam, Z. (2014). Oil palm root as biosorbent for heavy metals: biosorption, desorption and isothermal studies. *International Journal of ChemTech Research*, 6(1), pp. 163-177.

Björklund, K. & Li, L. (2015). Evaluation of low-cost materials for sorption of hydrophobic organic pollutants in stormwater. *Journal of Environmental Management*, 159(2015), pp. 106 - 114.

Boehm, H. P. (1994). Some aspects of the surface chemistry of carbon blacks and other carbons. *Carbon*, 32(5), pp. 759 - 769.

Bohart, G. S. & Adams, E. Q. (1920). Some aspects of the behavior of charcoal with respect to chlorine. *Journal of the American Chemical Society*, 42(3), pp. 523 - 544.

Booker, N. A., Cooney, E. L. & Priestley, A. J. (1996). Ammonia removal from sewage using natural Australian zeolite. *Water Science and Technology*, 34(9), pp. 17 - 24.

Boopathy, R., Karthikeyan, S., Mandal, A. B. & Sekaran, G. (2013). Adsorption of ammonium ion by coconut shell-activated carbon from aqueous solution: kinetic, isotherm, and thermodynamic studies. *Environmental Science and Pollution Research International*, 20(1), pp. 533 - 542.

- Bruker AXS. (2007). *S4 EXPLORER / S4 PIONEER X-ray Spectrometer*. Karlsruhe, GRMN.
- Burton, S. A. Q. & Watson-Craik, I. A. (1998). Ammonia and nitrogen fluxes in landfill sites: applicability to sustainable landfilling. *Waste Management and Research*, 16(1), pp. 41 - 53.
- Calace, N., Nardi, E., Petronio, B. M. & Pietroletti, M. (2002). Adsorption of phenols by papermill sludges. *Environmental Pollution*, 118(3), pp. 315 - 319.
- Calero, M., Hernáinz, F., Blázquez, G., Tenorio, G. & Martín-Lara, M. A. (2009). Study of Cr (III) biosorption in a fixed-bed column. *Journal of Hazardous Materials*, 171(1-3), pp. 886 - 893.
- Campbell, L. S. & Davies, B. E. (1995). Soil sorption of caesium modelled by the Langmuir and Freundlich isotherm equations. *Applied Geochemistry*, 10, pp. 715 - 723.
- Castilla, C. M. (2004). Adsorption of organic molecules from aqueous solutions on carbon materials. *Carbon*, 42(1), pp. 83 - 94.
- Cazetta, A. L., Vargas, A. M., Nogami, E. M., Kunita, M. H., Guilherme, M. R., Martins, A. C., Silva, T. L., Moraes, J. G. & Almeida, V. C. (2011). NaOHactivated carbon of high surface area produced from coconut shell: Kinetics and equilibrium studies from the methylene blue adsorption. *Chemical Engineering Journal*, 174(1), pp. 117 - 125.
- Chan, L. S., Cheung, W. H., Allen, S. J. & McKay, G. (2012). Error analysis of adsorption isotherm models for acid dyes onto bamboo derived activated carbon. *Chinese Journal of Chemical Engineering*, 20(3), pp. 535 - 542.
- Chen, H. H., Thirumavalavan, M., He, R. Z., Shih, Y. T. & Lee, J. F. (2017). Feasible preparation and characterization of tunable novel montmorillonite/block-copolymers based composites as potential dual adsorbent candidates. *Applied Clay Science*, 137, pp. 192 - 202.

- Chen, S., Yue, Q., Gao, B., Li, Q., Xu, X. & Fu, K. (2012). Adsorption of hexavalent chromium from aqueous solution by modified corn stalk: A fixed-bed column study. *Bioresource Technology*, 113(2012), pp. 114 - 120.
- Chien, S. H. & Clayton, W. R. (1980). Application of Elovich equation to the kinetics of phosphate release and sorption on soils. *Soil Science Society of America Journal*, 44(2), pp. 265 - 268.
- Chowdhury, Z. Z., Hamid, S. B. A. & Zain, S. M. (2015). Evaluating design parameters for breakthrough curve analysis and kinetics of fixed bed columns for adsorption studies of Cu (II) cations using lignocellulosic wastes. *BioResources*, 10(1), pp. 732 - 749.
- Christensen, T. H., Kjeldsen, P., Albrechtsen, H. J., Heron, G., Nielsen, P. H., Bjerg, P. L. & Holm, P. E. (1994). Attenuation of landfill leachate pollutants in aquifers. *Critical Reviews in Environmental Science and Technology*, 24(2), pp. 119 - 202.
- Chu, L. M., Cheung, K. C. & Wong, M. H. (1994). Variations in the chemical properties of landfill leachate. *Environmental Management*, 18(1), pp. 105 - 117.
- Çiftçi, T. D. (2017). Adsorptive properties of Fe₃O₄/Ni/Ni_xB nanocomposite coated nutshell for the removal of arsenic(iii) and arsenic (v) from waters. *Cogent Chemistry*, 3, pp. 1 – 15.
- Cincotti, A., Lai, N., Orrù, R. & Cao, G. (2001). Sardinian natural clinoptilolites for heavy metals and ammonium removal: experimental and modeling. *Chemical Engineering Journal*, 84(3), pp. 275 - 282.
- Contescu, A., Vass, M., Contescu, C., Putyera, K. & Schwarz, J. A. (1998). Acid buffering capacity of basic carbons revealed by their continuous pK distribution. *Carbon*, 36(3), pp. 247 - 258.
- Cooney, D. O. (1998). *Adsorption Design for Wastewater Treatment*. New York: CRC Press.

- Cyrus, J. S. & Reddy, G. B. (2011). Sorption and desorption of ammonium by zeolite: Batch and column studies. *Journal of Environmental Science and Health, Part A: Toxic/Hazardous Substances and Environmental Engineering*, 46(4), pp. 408 - 414.
- Dabwan, A. H. A., Egusa, K., Imai, D., Katsumata, H., Suzuki, T. & Kaneco, S. (2012). Utilization of paper sludge wastes for treatment of wastewater from food processing industries. *Pakistan Journal of Analytical & Environmental Chemistry*, 13(2), pp. 103 - 106.
- Dahlia, Z., Salmah, H. & Azlin, O. (2009). The effect of triethylene diamine on the properties of waste paper foam composites. *Journal of Physical Science*, 20(1), pp. 49 - 57.
- Dahri, M. K., Kooh, M. R. R. & Lim, L. B. L. (2014). Water remediation using low cost adsorbent walnut shell for removal of malachite green: Equilibrium, kinetics, thermodynamic and regeneration studies. *Journal of Environmental Chemical Engineering*, 2(3), pp. 1434 - 1444.
- Daud, Z. (2008). *Olahan Larut Lesapan Semi-Aerobik Tapak pelupusan Sanitari Pulau Burung Menggunakan Gabungan Kaedah Penggumpalan- Pengelompokan Dan Penurasan*. Universiti Sains Malaysia: Tesis Ph.D.
- Daud, Z., Abubakar, M. H., Kadir, A. A., Latiff, A. A. A., Awang, H., Halim, A. A. & Marto, A. (2017a). Adsorption studies of leachate on cockle shells. *International Journal of GEOMATE*, 12(29), pp. 46 - 52.
- Daud, Z., Abubakar, M. H., Abdul Kadir, A., Latiff, A. A. A., Awang, H., Halim, A. A. A. & Marto, A. (2017b). Batch study on COD and ammonia nitrogen removal using granular activated carbon and cockle shells. *International Journal of Engineering*, 30(7), pp. 937 – 944.
- Daud, Z., Abubakar, M. H., Kadir, A. A., Latiff, A. A. A., Awang, H., Halim, A. A. A. & Marto, A. (2017c). Leachate treatment optimization with granular activated carbon and cockle shells. *IJE TRANSACTIONS A: Basics*, 30(7), pp. 937 - 944.

- Daud, Z., Aziz, H. A. & Adlan, M. N. (2007). Treatment of semi-aerobic leachate by combined coagulation-flocculation and filtration method. *Journal of Environmental Research and Development*, 2(2), pp. 101 - 110.
- Daud, Z., Ibrahim, F. N. D., Latiff, A. A. A., Ridzuan, M. B., Ahmad, Z., Awang, H. & Marton, A. (2016). Ammoniacal nitrogen and COD removal using zeolite feldspar mineral composite adsorbent. *International Journal of Integrated Engineering*, 8(3), pp. 9 - 12.
- Demir, A., Günay, A. & Debik, E. (2002). Ammonium removal from aqueous solution by ion-exchange using packed bed natural zeolite. *Water S. A.*, 28(3), pp. 329 - 336.
- Deurer, M., Muller, K., Dijssel, C. V. D., Mason, K., Carter, J. & Clothier, B. E. (2011). Is soil water repellency a function of soil order and proneness to drought? A survey of soils under pasture in the North Island of New Zealand. *European Journal of Soil Science*, 62(6), pp. 765 - 779.
- Devi, P. & Saroha, A. K. (2014). Synthesis of the magnetic biochar composites for use as an adsorbent for the removal of pentachlorophenol from the effluent. *Bioresource Technology*, 169(2014), pp. 525 - 531.
- Devi, P. & Saroha, A. K. (2015). Effect of pyrolysis temperature on polycyclic aromatic hydrocarbons toxicity and sorption behavior of biochars prepared by pyrolysis of paper mill effluent treatment sludge. *Bioresource Technology*, 192, pp. 312 - 320.
- Dias, J. M., Alvim-Ferraz, M. C. M., Almeida, M. F., Utrilla, J. R. & Polo, M. S. (2007). Waste materials for activated carbon preparation and its use in aqueous-phase treatment: A review. *Journal of Environmental Management*, 85(4), pp. 833 - 846.
- Dias, N. C., Steiner, P. A. & Braga, M. C. B. (2015). Characterization and modification of a clay mineral used in adsorption tests. *Journal of Minerals and Materials Characterization and Engineering*, 3(4), pp. 277 - 288.
- Drioli, E., Criscuoli, A. & Curcio, E. (2011). *Membrane Contactors: Fundamentals, Applications and Potentialities*. Edisi ke-11. Elsevier Science.

- Du, Q., Liu, S., Cao, Z. & Wang, Y. (2005). Ammonia removal from aqueous solution using natural Chinese clinoptilolite. *Separation and Purification Technology*, 44(3), pp. 229 - 234.
- Dwivedi, C. P., Sahu, J., Mohanty, C., Mohan, B. R., & Meikap, B. (2008). Column performance of granular activated carbon packed bed for Pb (II) removal. *Journal of Hazardous Materials*, 156(1), pp. 596 - 603.
- Ehrig, H. J. (1988). Water and element balances in landfills. In *Lecture Notes in Earth Sciences*, (P. Baccini, ed.), Berlin: Springer-Verlag, pp. 83 - 115.
- Ekpete, O. A. & Horsfall, M. JNR. (2011). Preparation and characterization of activated carbon derived from fluted pumpkin stem waste (*Telfairia occidentalis* Hook F). *Research Journal of Chemical Sciences*, 1(3), pp. 10 - 17.
- Ekpete, O. A., Horsfall, M. JNR. & Tarawou, T. (2011). Evaluation of activated carbon from fluted pumpkin stem waste for fenol and chlorofenol adsorption in a fixed-bed micro-column. *Journal Applied Science and Environmental Management*, 15(1), pp. 141 - 146.
- El-Fadel, M., Bou-Zeid, E., Chahine, W. & Alayli, B. (2002). Temporal variation of leachate quality from pre-sorted and baled municipal solid waste with high organic and moisture content. *Waste Management*, 22(3), pp. 269 - 282.
- El-Fadel, M., Findikakis, A. N. & Leckie, J. O. (1997). Environmental impacts of solid waste landfilling. *Journal of Environmental Management*, 50(1), pp. 1 - 25.
- Emadi, H., Nezhad, J. E. & Pourbagher, H. (2001). In vitro comparison of zeolite (clinoptilolite) and activated carbon as ammonia absorbants in fish culture. *Naga, the ICLARM Quarterly*, 24(1/2), pp. 18 - 20.
- Environmental Protection Agency (2000). Landfill manuals and Landfill site design (EPA) [Online] Capaian maklumat pada 14 Disember 2014 dari https://www.epa.ie/pubs/advice/waste/waste/EPA_landfill_site_design_guide.pdf

- Fettig, J. (1999). Removal of humic substances by adsorption/ion exchange. *Water Science and Technology*, 40(9), pp. 171 - 182.
- Foo, K. Y. & Hameed, B. H. (2009). An overview of landfill leachate treatment via activated carbon adsorption process. *Journal of Hazardous Materials*, 171(1-3), pp. 54 - 60.
- Foo, K. Y. & Hameed, B. H. (2010). Insights into the modeling of adsorption isotherm systems. *Chemical Engineering Journal*, 156(1), pp. 2 - 10.
- Foo, K. Y. & Hameed, B. H. (2011). The environmental applications of activated carbon/zeolite composite materials. *Advances in Colloid and Interface Science*, 162(1-2), pp. 22 - 28.
- Foo, K. Y., Lee, L. K. & Hameed, B. H. (2013). Batch adsorption of semi-aerobic landfill leachate by granular activated carbon prepared by microwave heating. *Chemical Engineering Journal*, 222(2013), pp. 259 - 264.
- Foul, A. A., Aziz, H. A., Isa, M. H. & Hung, Y. -T. (2009). Primary treatment of anaerobic landfill leachate using activated carbon and limestone: Batch and column studies. *International Journal of Environment and Waste Management*, 4(3/4), pp. 282 - 298.
- Freundlich, H. M. F. (1906). Over the adsorption in solution. *Journal of Physical Chemistry*, 57, pp. 385 - 470.
- Frías, M., Rodríguez, O. & Sánchez de Rojas, M. I. (2004). Paper sludge, an environmentally sound alternative source of MK-based cementitious materials. A review. *Construction and Building Materials*, 74, pp. 37 – 48.
- García, R., Vigil de la Villa, R., Vegas, I., Frías, M. & Sánchez de Rojas, M. I. (2007). The pozzolanic properties of paper sludge waste. *Construction and Building Materials*, 22, pp. 1484 – 1490.

- Gao, J., Kong, D., Wang, Y., Wu, J., Sun, S. & Xu, P. (2013). Production of mesoporous activated carbon from tea fruit peel residues and its evaluation of methylene blue removal from aqueous solutions. *BioResources*, 8(2), pp. 2145 - 2160.
- Gao, Z., Bandosz, T. J., Zhao, Z., Han, M. & Qiu, J. (2009). Investigation of factors affecting adsorption of transition metals on oxidized carbon nanotubes. *Journal of Hazardous Materials*, 167, pp. 357 – 365.
- Gluth, J. G. G., Lehmann, C., Rübner, K. & Kühne, H. (2013). Reaction products and strength development of wastepaper sludge ash and the influence of alkalis. *Cement & Concrete Composites*, 45, pp. 82 – 88.
- Halim, A. A. (2008). *Olahan Larut Lesapan Semi-Aerobik Menggunakan Penjerap Komposit Berasaskan Bahan Mineral dan Organik*. Universiti Sains Malaysia: Tesis Ph.D.
- Halim, A. A. & Ahmad, M. F. (2013). Isoterma dan kinetik penjerapan boron oleh batu kapur sebagai penjerap berkos rendah. *Sains Malaysiana*, 42(12), pp. 1689 - 1696.
- Halim, A. A., Abidin, N. N. Z., Awang, N., Ithnin, A., Othman, M. S. & Wahab, M. I. (2011). Ammonia and COD removal from synthetic leachate using rice husk composite adsorbent. *Journal of Urban and Environmental Engineering*, 5(1), pp. 24 - 31.
- Halim, A. A., Aziz, H. A., Megat Johari, M. A. & Ariffin, K. S. (2006). Landfill leachate treatment using combination of hydrophobic-hydrophilic and low cost adsorption materials as a single media. *Proceeding of 1st Civil Engineering Colloquium (CEC'06)*. School of Civil Engineering, USM, Nibong Tebal, Penang, Malaysia.
- Halim, A. A., Aziz, H. A., Megat Johari, M. A. & Ariffin, K. S. (2010). Comparison study of ammonia and COD adsorption on zeolite, activated carbon and composite materials in landfill leachate treatment. *Desalination*, 262(1-3), pp. 31 - 35.
- Halim, A. A., Aziz, H. A., Megat Johari, M. A., Ariffin, K. S. & Hung, Y. T. (2009). Removal of ammoniacal nitrogen and COD from semi-aerobic landfill leachate

- using low-cost activated carbon-zeolite composite adsorbent. *International Journal of Environment and Waste Management*, 4(3/4), pp. 399 - 411.
- Halim, A. A., Aziz, H. A., Megat Johari, M. A., Ariffin, K. S. & Bashir, M. J. K. (2012). Semi-aerobic landfill leachate treatment using carbon-minerals composite adsorbent. *Environmental Engineering Science*, 29(5), pp. 306 - 312.
- Halim, A. A., Han, K. K. & Hanafiah, M. M. (2015). Removal of methylene blue from dye wastewater using river sand by adsorption. *Nature Environment and Pollution Technology*, 14(1), pp. 89 - 94.
- Halimoon, N. & Yin, R. G. S. (2010). Removal of heavy metals from textile wastewater using zeolite. *Environment Asia 3 (special issue)*, pp. 124 - 130.
- Hallbourg, R. R., Delfino, J. J. & Miller, W. L. (1992). Organic priority pollutants in groundwater and surface water at three landfills in north central Florida. *Water Air and Soil Pollution*, 65(3), pp. 307 - 322.
- Haseena, P. V., Padmavathy, K. S., Krishnan, P. R. & Madhu, G. (2016). Adsorption of ammonium nitrogen from aqueous systems using chitosan-bentonite film composite. *Procedia Technology*, 24(2016), pp. 733 - 740.
- Hasfalina, C. M., Maryam, R. Z., Luqman, C. A. & Rashid, M. (2012). Adsorption of copper (II) from aqueous medium in fixed-bed column by kenaf fibres. *APCBEE Procedia*, 3(2012), pp. 255 - 263.
- Hlavay, J., Vigh, G. Y., Olaszi, V. & Inczedi, J. (1982). Investigation on natural Hungarian zeolite for ammonia removal. *Water Research*, 16(4), pp. 417 - 420.
- Hofman, M. & Pietrzak, R. (2012). NO₂ removal by adsorbents prepared from waste paper sludge. *Chemical Engineering Journal*, 183, pp. 278 - 283.
- Hojamberdiev, M., Kameshima, Y., Nakajima, A., Okada, K. & Kadirova, Z. (2008). Preparation and sorption properties of materials from paper sludge. *Journal of Hazardous Materials*, 151(2-3), pp. 710 - 719.

- Hor, K. Y., Chee, J. M. C., Chong, M. N., Jin, B., Saint, C., Poh, P. E. & Aryal, R. (2016). Evaluation of physicochemical methods in enhancing the adsorption performance of natural zeolite as low-cost adsorbent of methylene blue dye from wastewater. *Journal of Cleaner Production*, 118, pp. 197 - 209.
- Ho, Y. S. & McKay, G. (2002). Application of kinetic models to the sorption of copper (II) on to peat. *Adsorption Science and Technology*, 20(8), pp. 797 - 815.
- Ho, Y. S., Porter, F. & McKay, G. (2002). Equilibrium isotherm studies for the sorption of divalent metal ions onto peat: copper, nickel and lead single component systems. *Water, Air and Soil Pollution*, 141(1), pp. 1 - 33.
- Huang, C. C., Li, H. S. & Chen, C. H. (2008). Effect of surface acidic oxides of activated carbon on adsorption of ammonia. *Journal of Hazardous Materials*, 159(2-3), pp. 523 - 527.
- Huang, D. J. & Leu, T. S. (2013). Fabrication of high wettability gradient on copper substrate. *Applied Surface Science*, 280(2013), pp. 25 - 32.
- Iglauer, S., Salamah, A., Sarmadivaleh, M., Liu, K. & Phan, C. (2014). Contamination of silica surfaces: impact on water-CO₂-quartz and glass contact angle measurements. *International Journal of Greenhouse Gas Control*, 22, pp. 325 - 328.
- Igwe, J. C. & Abia, A. A. (2007). Adsorption kinetics and intraparticulate diffusivities for bioremediation of Co (II), Fe (II) and Cu (II) ions from waste water using modified and unmodified maize cob. *International Journal of Physical Sciences*, 2(5), pp. 119 - 127.
- Inanc, B., Calli, B. & Saatci, A. (2000). Characterization and anaerobic treatment of the sanitary landfill leachate in Istanbul. *Water Science and Technology*, 41(3), pp. 223 - 230.

- Inglezakis, V. J., Zorpas, A. A., Loizidou, M. D. & Grigoropoulou, H. P. (2005). The effect of competitive cations and anions on ion exchange of heavy metals. *Separation and Purification Technology*, 46(3), pp. 202 - 207.
- Isa, M. H., Lang, L. S., Asaari, F. A. H., Aziz, H. A., Ramli, N. A. & Dhas, J. P. A. (2007). Low cost removal of disperse dyes from aqueous solution using palm ash. *Dyes and Pigments*, 74(2), pp. 446 - 453.
- Itodo, A.U., Abdulrahman, F.W., Hassan, L.G., Maigandi, S.A. & Itodo, H.U. (2010). Application of methylene blue and iodine adsorption in the measurement of specific surface area by four acid and salt treated activated carbons. *New York Science Journal*, 3(5), pp. 25 - 33.
- Jabatan Alam Sekitar Malaysia (2010). [Online] Capaian maklumat pada 18 Ogos 2016 dari <http://www.doe.gov.my/eia/wp-content/uploads/2012/03/A-Guide-For-Investors1.pdf>
- Jackson, M. J. & Line, M. A. (1997). Organic composition of a pulp and paper mill sludge determined by FTIR, ¹³C CP MAS NMR, and chemical extraction techniques. *Journal of Agricultural and Food Chemistry*, 45(6), pp. 2354 - 2358.
- Jha, V. K., Kameshima, Y., Nakajima, A., Okada, K. & Mackenzie, K. J. D. (2006). Multifunctional uptake behaviour of material prepared by calcining waste paper sludge. *Journal of Environmental Science and Health*, 41(4), pp. 703 - 719.
- Jokela, J. P. Y., Kettunan, R. H., Sormunen, K. M. & Rintala, J. A. (2002). Biological nitrogen removal from municipal landfill leachate: low-cost nitrification in biofilters and laboratory scale in-situ denitrification. *Water Research*, 36(16), pp. 4079 - 4087.
- Jorgensen, T. C. & Weatherley, L. R. (2003). Ammonia removal from wastewater by ion exchange in the presence of organic contaminants. *Water Research*, 37(8), pp. 1723 - 1728.

- Kadirvelu, K., Kavipriya, M., Karthika, C., Radhika, M., Vennilamani, N. & Pattabhi, S. (2003). Utilization of various agricultural wastes for activated carbon preparation and application for the removal of dyes and metal ions from aqueous solutions. *Bioresource Technology*, 87(1), pp. 129 - 132.
- Kalmykova, Y., Moona, N., Stromvall, A. M. & Bjorklund, K. (2014). Sorption and degradation of petroleum hydrocarbons, polycyclic aromatic hydrocarbons, alkylphenols, bisphenol A and phthalates in landfill leachate using sand, activated carbon and peat filters. *Water Research*, 1(56), pp. 246 - 257.
- Kamaruddin, M. A., Yusoff, M. S. & Ahmad, M. A. (2011). Optimization of durian peel based activated carbon preparation conditions for ammoniacal nitrogen removal from semi-aerobic landfill leachate. *Journal of Scientific & Industrial Research*, 70(7), pp. 554 – 560.
- Kamaruddin, M. A., Yusoff, M. S., Aziz, H. A. & Rui, L. M. (2012). Influence of impregnation ratio on chemically modified silica sand for heavy metals removal from stabilized landfill leachate. *International Journal of Environmental Protection*, 2(3), pp. 15 - 22.
- Kamaruddin, M. A., Yusoff, M. S., Aziz, H. A. & Ismail, M. N. (2013). Preparation and characterization of composite embedded clinoptilolite for the removal of color and lead from textile waste water. *International Journal of Scientific Research in Inventions and New Ideas (IJSRIN)*, 1(2), pp. 37 - 47.
- Kamaruddin, M. A., Yusoff, M. S., Aziz, H. A. & Hung, Y-T. (2015). Sustainable treatment of landfill leachate. *Applied Water Science*, 5(2), pp. 113 - 126.
- Kanawade, S. M. (2016). Treatment on synthetic effluent by using limestone and granular activated carbon for removal of ammoniacal nitrogen. *International Journal of Advanced Research and Development*, 1(3), pp. 55 - 63.
- Karadag, D., Tok, S., Akgul, E., Turan, M., Ozturk, M. & Demir, A. (2008). Ammonium removal from sanitary landfill leachate using natural Gordes clinoptilolite. *Journal of Hazardous Materials*, 153(1-2), pp. 60 - 66.

- Kargi, F. & Pamukoglu, M. Y. (2004). Adsorbent supplemented biological treatment of pre-treated landfill leachate by fed-batch operation. *Bioresource Technology*, 94(3), pp. 285 - 291.
- Kaur, K, Mor, S. & Ravindra, K. (2016). Removal of chemical oxygen demand from landfill leachate using cow-dung ash as a low-cost adsorbent. *Journal of Colloid and Interface Science*, 469(1), pp. 338 - 343.
- Kehinde, F. O. & Aziz, H. A. (2015). Clinoptilolite assisted persulfate decolorization of a raw textile wastewater. *American Chemical Science Journal*, 8(3), pp. 1 - 7.
- Khamidun, M. H. & Fulazzaky, M. A. (2015). Adsorption of phosphate from synthetic solution onto the limestone in a plug-flow column. *International Journal of Advanced and Applied Sciences*, 2(12), pp. 7 - 13.
- Kibami, D., Pongener, C., Rao, K. S. & Sinha, D. (2014). Preparation and characterization of activated carbon from Fagopyrum esculentum Moench by HNO_3 and H_3PO_4 chemical activation. *Der Chemica Sinica*, 5(4), pp. 46 - 55.
- Kinney, T. J., Masiello, C. A., Dugan, B., Hockaday, W. C., Dean, M. R., Zygourakis, K. & Barnes, R. T. (2012). Hydrologic properties of biochars produced at different temperatures. *Biomass and Bioenergy*, 41, pp. 34 - 43.
- Kithome, M., Paul, J. W., Lavkulich, L. M. & Bomke, A. A. (1998). Kinetics of ammonium adsorption and desorption by the natural zeolite clinoptilolite, *Soil Science Society of America Journal*, 62(3), pp. 622 - 629.
- Kjeldsen, P., Barlaz, M. A., Rooker, A. P., Baun, A., Ledin, A. & Christensen, T. H. (2002). Present and long-term composition of MSW landfill leachate: A review, *Critical Reviews in Environmental Science and Technology*, 32(4), pp. 297 - 336.
- Kořenková, L. & Urik, M. (2015). Basic soil properties as a factor controlling the occurrence and intensity of water repellency in rankers of the White Carpathians. *Folia Forestalia Polonica, series A*, 57(3), pp. 129 - 137.

- Krishni, R. R., Foo, K. Y. & Hameed, B. H. (2014). Adsorption of methylene blue onto papaya leaves: comparison of linear and nonlinear isotherm analysis. *Desalination and Water Treatment*, 52(34-36), pp. 6712 - 6719.
- Kristensen, K. P. (2007). *Biodegradation of Xenobiotic Organic Compounds in Wastewater Treatment Plants*. Institute of Environment & Resources Technical University of Denmark: Tesis Ph.D.
- Kruempelbeck, I. & Ehrig, H. J. (1999). Long term behavior of municipal solid waste landfills in Germany. Sardinia 99, Seventh International Waste Management and Landfill Symposium, October 4 - 7, 1, 27 - 36.
- Kucić, D., Cosić, I., Vuković, M. & Briski, F. (2013). Sorption kinetic studies of ammonium from aqueous solution on different inorganic and organic media. *Acta Chimica Slovenica*, 60(1), pp. 109 - 119.
- Kulikowska, D. & Klimiuk, E. (2008). The effect of landfill age on municipal leachate composition. *Bioresource Technology*, 99(3), pp. 5981 - 5985.
- Kulkarni, S. J. (2013). Removal of organic matter from domestic waste water by adsorption. *International Journal of Science, Engineering and Technology Research (IJSETR)*, 2(10), pp. 1836 - 1839.
- Kundu, S., Kavalakatt, S. S., Pal, A., Ghosh, S. K., Mandal, M. & Pal, T. (2004). Removal of arsenic using hardened paste of portland cement: batch adsorption and column study. *Water Research*, 38(17), pp. 3780 - 3790.
- Kurniawan, T. A., Lo, W. -H. & Chan, G. Y. S. (2006). Physico-chemical treatments for removal of recalcitrant contaminants from landfill leachate. *Journal of Hazardous Materials*, 129(1-3), pp. 80 - 100.
- Lafi, R., Fradj, A. B., Hafiane, A. & Hameed, B. H. (2014). Coffee waste as potential adsorbent for the removal of basic dyes from aqueous solution. *Korean Journal of Chemical Engineering*, 31(12), pp. 2198 - 2206.

- Lakhera, S. K., Sree, H. A. & Suman, S. (2015). Synthesis and characterization of 13x zeolite/ activated carbon composite. *International Journal of ChemTech Research*, 7(3), pp. 1364 - 1368.
- Langmuir, I. (1918). The adsorption of gases on plane surface of glass, mica and platinum. *Journal of American Chemical Society*, 40(9), pp. 1361 - 1403.
- Lebedynets, M., Sprynskyy, M., Sakhnyuk, I., Zbytyniewski, R., Golembiewski, R. & Buszewski, B. (2004). Adsorption of ammonium ions onto a natural zeolite: Transcarpathian clinoptilolite. *Adsorption Science & Technology*, 22(9), pp. 731 - 741.
- Leboda, R. (1992). Effect of silica gel quantity on the course of hydrothermal treatment in an autoclave. *Materials Chemistry and Physics*, 31(3), pp. 243 - 255.
- Leboda, R. (1993). Carbon-mineral adsorbents-new type of sorbents part II. Surface properties and methods of their modification. *Materials Chemistry and Physics*, 34(2), pp. 123 - 141.
- Lee, A. H., Nikraz, H. & Hung, Y. T. (2010). Influence of waste age on landfill leachate quality. *International Journal of Environmental Science and Development*, 1(4), pp. 347 - 350.
- LeVan, M. D. & Vermeulen, T. (1981). Binary Langmuir and Freundlich isotherms for ideal adsorbed solution. *The Journal of Physical Chemistry*, 85(22), pp. 3247 - 3250.
- Lei, L., Xiaojuan, L. & Zhang, X. (2008). Ammonium removal from aqueous solutions using microwave-treated natural Chinese zeolite. *Separation and Purification Technology*, 58(3), pp. 359 - 366.
- Liaw, C. T., Chang, H. L., Hsu, W. C. & Huang, C. R. (1998). A novel method to reuse paper sludge and co-generation ashes from paper mill. *Journal of Hazardous Materials*, 58(1-3), pp. 93 - 102.

- Likon, M., Cernec, F., Svegl, F., Saarela, J. & Zimmie, T. F. (2011). Papermill industrial waste as a sustainable source for high efficiency absorbent production. *Waste Management*, 31(6), pp. 1350 - 1356.
- Liu, Z., Wu, W., Shi, P., Guo, J. & Cheng, J. (2015). Characterization of dissolved organic matter in landfill leachate during the combined treatment process of air stripping, Fenton, SBR and coagulation. *Waste Management*, 41, pp. 111 - 118.
- Li, W., Zhou, Q. & Hua, T. (2010). Removal of organic matter from landfill leachate by advanced oxidation processes: A review. *International Journal of Chemical Engineering*, 1-10. [Online] Capaian maklumat pada 20 Januari 2016 dari <http://dx.doi.org/10.1155/2010/270532>
- Lim, C. K., Bay, H. H., Neoh, C. H., Aris, A., Majid, Z. A. & Ibrahim, Z. (2013). Application of zeolite-activated carbon macrocomposite for the adsorption of Acid Orange 7: isotherm, kinetic and thermodynamic studies. *Environmental Science and Pollution Research*, 20(10), pp. 7243 - 7255.
- Lim, C. K., Seow, T. W., Neoh, C. H., Nor, M. H. M., Ibrahim, Z., Ware, I. & Sarip, S. H. M. (2016). Treatment of landfill leachate using ASBR combined with zeolite adsorption technology. *3 Biotech*, 6(195), pp. 1 - 6.
- Lin, S. H. & Wu, C. L. (1996). Removal of nitrogenous compounds from aqueous solution by ozonation and ion exchange. *Water Research*, 30(8), pp. 1851 - 1857.
- Lin, W., Tadaï, O., Takahashi, M., Sato, D., Hirose, T., Tanikawa, W., Hamada, Y. & Hatakeda, K. (2015). An experimental study on measurement methods of bulk density and porosity of rock samples. *Journal of Geoscience and Environment Protection*, 3, pp. 72 - 79.
- Lorenzen, L., Van Deventer, J. S. J. & Landi, W. M. (1995). Factors affecting the mechanism of the adsorption of arsenic species on activated carbon. *Minerals Engineering*, 8(4/5), pp. 557 - 569.

- Lupul, I., Yperman, J., Carleer, R. & Gryglewicz, G. (2015). Adsorption of atrazine on hemp stem-based activated carbons with different surface chemistry. *Adsorption*, 21(6-7), pp. 489 - 498.
- Manocha, S. M. (2003). Porous carbons. *Sadhana*, 28(1-2), pp. 335 - 348.
- Manu, B. & Chaudhari, S. (2002). Anaerobic decolorization of simulated textile wastewater containing azo dyes. *Bioresource Technology*, 82(3), pp. 225 - 231.
- Marsh, H. & Rodriguez-Reinoso, F. (2006). *Activated Carbon*. Amsterdam: Elsevier Science & Technology Books.
- Mashal, A., Dahrieh, J. A., Ahmad, A. A., Oyedele, L., Haimour, N., Ali, A. A. H. & Rooney, D. (2014). Fixed-bed study of ammonia removal from aqueous solutions using natural zeolite. *World Journal of Science, Technology and Sustainable Development*, 11(2), pp. 144 - 158.
- McKay, G. & Bino, M. J. (1990). Fixed bed adsorption for the removal of pollutants from water. *Environmental Pollution*, 66(1), pp. 33 - 53.
- Mendez, A., Barigga, S., Fidalgo, J. M. & Gasco, G. (2009). Adsorbent materials from paper industry waste materials and their use in Cu (II) removal from water. *Journal of Hazardous Materials*, 165(1-3), pp. 736 - 743.
- Mendez, A., Paz-Ferreiro, J., Araujo, F. & Gasco, G. (2014). Biochar from pyrolysis of deinking paper sludge and its use in the treatment of a nickel polluted soil. *Journal of Analytical and Applied Pyrolysis*, 107, pp. 46 - 52.
- Metcalf & Eddy, INC. (2003). *Wastewater Engineering: Treatment and Reuse* (McGraw Hill series in civil and environmental engineering). Edisi Keempat. New York: McGraw-Hill.
- Meyer, M. L. & Bloom, P. R. (1997). Boric and silicate acid adsorption and desorption by a humic acid. *J. Environ. Qual.*, 26(1).

- Mihelcic, R. J. & Zimmerman, B. J. (2014). *Environmental Engineering: Fundamentals, Sustainability, Design*. Edisi Kedua. Wiley.
- Misaelides, P. (2011). Application of natural zeolites in environmental remediation: A short review. *Microporous and Mesoporous Materials*, 144(1-3), pp. 15 - 18.
- Mohammadizaroun, M. & Yusoff, M. S. (2014). Review on landfill leachate treatment using physical-chemical techniques: their performance and limitations. *International Journal of Current Life Sciences*, 4(12), pp. 12068 - 12074.
- Moideen, S. N. F., Din, M. F. M., Rezanian, S., Ponraj, M., Rahman, A. A., Pei, L. W., Ismail, Z., Taib, S. M., Li, Y. Y. & Komori, D. (2017). Dual phase role of composite adsorbents made from cockleshell and natural zeolite in treating river water. *Journal of King Saud University - Science*, doi: <http://dx.doi.org/10.1016/j.jksus.2017.06.001>
- Mojiri, A. (2011). Review on membrane bioreactor, ion exchange and adsorption methods for landfill leachate treatment. *Australian Journal of Basic and Applied Sciences*, 5(12), pp. 1365 - 1370.
- Mojiri, A., Aziz, H. A., Qamaruz Zaman, N. & Aziz, S. Q. (2012a). A review on anaerobic digestion, bio-reactor and nitrogen removal from wastewater and landfill leachate by bio-reactor. *Advances in Environmental Biology*, 6(7), pp. 2143 - 2150.
- Mojiri, A., Aziz, H. A., Aziz, S. Q. & Qamaruz Zaman, N. (2012b). Review on municipal landfill leachate and sequencing batch reactor (SBR) technique. *Archives Des Sciences*, 65(7), pp. 22 - 31.
- Mojiri, A., Aziz, H. A. & Aziz, S. Q. (2013). Trends in physical-chemical methods for landfill leachate treatment. *International Journal of Scientific Research in Environmental Sciences (IJSRES)*, 1(2), pp. 16 - 25.
- Mojiri, A., Aziz, H. A., Zaman, N. Q., Aziz, S. Q. & Zahed, M. A. (2014). Powdered ZELIAC augmented sequencing batch reactors (SBR) process for co-treatment of

- landfill leachate and domestic wastewater. *Journal of Environmental Management*, 139, pp 1 - 14.
- Mojiri, A., Aziz, H. A., Zaman, N. Q., Aziz, S. Q. dan Zahed, M. A. (2016). Metals removal from municipal landfill leachate and wastewater using adsorbents combined with biological method. *Desalination and Water Treatment*, 57(6), pp. 2819 - 2833.
- Montalvo, S., Guerrero, L., Borja, R., Sánchez, E., Milán, Z., Cortés, I. & Delala Rubia, M. A. (2012). Application of natural zeolites in anaerobic digestion processes: a review. *Applied Clay Science*, 58, pp. 125 - 133.
- Mor, S., Ravindra, K., Dahiya, R. P. & Chandra, A. (2006). Leachate characterization and assessment of groundwater pollution near municipal solid waste landfill site. *Environmental Monitoring and Assessment*, 118, pp. 435 – 456.
- Motling, S, Mukherjee, S. N. & Dutta, A. (2014). Removal of phenolic compound and cod from landfill leachate by commercial activated carbon. *International Journal of Engineering Research & Technology*, 3(2), pp. 2880 - 2888.
- Nguyen, M. L. & Tanner, C. C. (1998). Ammonium removal from wastewaters using natural New Zealand zeolites. *New Zealand Journal of Agricultural Research*, 41(3), pp. 427 - 446.
- Nguyen, P. T., Nguyen, T. A., Bhandari, B. & Prakash, S. (2016). Comparison of solid substrates to differentiate the lubrication property of dairy fluids by tribological measurement. *Journal of Food Engineering*, 185, pp. 1 - 8.
- Nishikawa, M., Tanaka, K., Uetake, M., Enoda, M., Kawamura, Y. & Okuno, K. (1995). Adsorption isotherm and separation factor for multicomponent hydrogen isotopes in cryosorption method for recovery of tritium from blanket sweep gas. *Fusion Technology*, 28, pp. 711 – 716.
- Nor Nazrieza, M. S., Siti Rohana, M. Y., Subramaniam, K., Hazilia, H. & Amir Herberd, A. (2015). Characterization of leachate from Panchang Bedena landfill, Batang

- Padang landfill and Matang landfill: A comparative study. *Malaysian Journal of Science*, 34(1), pp. 69 - 77.
- Ntampou, X., Zouboulis, A. I. & Samaras, P. (2006). Appropriate combination of physicochemical methods (coagulation/flocculation and ozonation) for the efficient treatment of landfill leachates. *Chemosphere*, 62(5), pp. 722 - 730.
- Nurazim, I., Hamidi, A. A. & Mohd, S. Y. (2017). Adsorption of UV₂₅₄ in Kerian River water onto ZeliacTM: Analysis using linear and non-linear forms of isotherm models. *Global NEST Journal*, 19(1), pp. 74 - 81.
- Oboh, I. O., Aluyor, E. O. & Audu, T. O. K. (2013). Second-order kinetic model for the adsorption of divalent metal ions on *Sida acuta* leaves. *International Journal of Physical Sciences*, 8(34), pp. 1722 - 1728.
- Okada, K., Ono, Y., Kameshima, Y., Nakajima, A. & MacKenzie, K. J. D. (2007). Simultaneous uptake of ammonium and phosphate ions by compounds prepared from paper sludge ash. *Journal of Hazardous Materials*, 141(3), pp. 622 - 629.
- Okolo, B., Park, C. & Keane, M. A. (2000). Interaction of phenol and chlorophenols with activated carbon and synthetic zeolites in aqueous media. *Journal of Colloid and Interface Science*, 226(2), pp. 308 - 317.
- Oloibiri, V., Coninck, S. D., Chys, M., Demeestere, K. & Hulle, S. W. H. V. (2017). Characterisation of landfill leachate by EEM-PARAFAC-SOM during physical-chemical treatment by coagulation-flocculation, activated carbon adsorption and ion exchange. *Chemosphere*, 186, pp. 873 - 883.
- Olorunfemi, I. E., Ogunrinde, T. A. & Fasinmirin, J. T. (2014). Soil hydrophobicity: an overview. *Journal of Scientific Research & Reports*, 3(8), pp. 1003 - 1037.
- Öman, C. B. & Junestedt, C. (2008). Chemical characterization of landfill leachates – 400 parameters and compounds. *Waste Management*, 28(10), pp. 1876 - 1891.
- Ono, Y. & Yashima, T. (Eds.) (2000). *Science and Engineering of Zeolite*. Kodansha-Tokyo: Scientific Publishing.

- Othman, E., Yusoff, M. S., Aziz, H. A., Adlan, M. N., Bashir, M. J. K. & Hung, Y. T. (2010). The effectiveness of silica sand in semi-aerobic stabilized landfill leachate treatment. *Water*, 2(4), pp. 904 - 915.
- Park, S-. J. & Kim, B-. J. (2005). Ammonia removal of activated carbon fibers produced by oxyfluorination. *Journal of Colloid and Interface Science*, 291(2), pp. 597 - 599.
- Perego, C., Bagatin, R., Tagliabue, M. & Vignola, R. (2013). Zeolites and related mesoporous materials for multi-talented environmental solutions. *Microporous and Mesoporous Materials*, 166, pp. 37 - 49.
- Pizarro, C., Rubio, M. A., Escudey, M., Albornoz, M. F., Munoz, D., Denardin, J. & Fabris, J. D. (2015). Nanomagnetite-zeolite composites in the removal arsenate from aqueous systems. *Journal of the Brazilian Chemical Society*, 26(9), pp. 1887 - 1896.
- Qiu, H., Lu, L., Pan, B-. C., Zhang, Q-. J., Zhang, W-. M. & Zhang, Q-. X. (2009). Critical review in adsorption kinetic models. *Journal of Zhejiang University SCIENCE A*, 10(5), pp. 716 - 724.
- Rafizul, I. M. & Alamgir, M. (2012). Characterization and tropical seasonal variation of leachate: Results from landfill lysimeter studied. *Waste Management*, 32(11), pp. 2080 - 2095.
- Rahmani, A. R., Mahvi, A. H., Mesdaghinia, A. R. & Nasser, S. (2004). Investigation of ammonia removal from polluted waters by Clinoptilolite zeolite. *International Journal of Environmental Science & Technology*, 1(2), pp. 125 - 133.
- Rahmani, A. R., Samadi, M. T. & Ehsani, H. R. (2009). Investigation of clinoptilolite natural zeolite regeneration by air stripping followed by ion exchange for removal of ammonium from aqueous solutions. *Iranian Journal of Environmental Health Science & Engineering*, 6(3), pp. 167 - 172.

- Reinhart, D. R. & Townsend, T. G. (1997). *Landfill Bioreactor Design and Operation*. USA: Lewis publisher.
- Renou, S., Givaudan, J. G., Poulain, S., Dirassouyan, F. & Moulin, P. (2008). Landfill leachate treatment: review and opportunity. *Journal of Hazardous Materials*, 150(3), pp. 468 - 493.
- Robinson, H. D. (1995). A review of the composition of leachates from domestic wastes in landfill sites. A Report for the UK Dep. Of the Environment, 2 Marsham Str., London SW1P 3EB, UK. 5 parts (A-E). Reference: DE0918A/FRI.
- Rodriguez, J., Castrillon, L., Maranon, E., Sastre, H. & Fernandez, E. (2004). Removal of non-biodegradable organic matter from landfill leachate by adsorption. *Water Research*, 38(14-15), pp. 3297 - 3303.
- Rosli, M. A. (2017). *Potensi Tanah Gambut, Batu Kapur, Zeolit dan Karbon Teraktif Sebagai Penjerap Komposit Untuk Merawat Larut Resapan*. Universiti Tun Hussein Onn Malaysia: Tesis Ph.D.
- Rouf, S. & Nagapadma, M. (2015). Modeling of fixed bed column studies for adsorption of azo dye on chitosan impregnated with a cationic surfactant. *International Journal of Scientific & Engineering Research*, 6(2), pp. 538 - 544.
- Rout, P. R., Dash, R. R. & Bhunia, P. (2014). Modelling and packed bed column studies on adsorptive removal of phosphate from aqueous solutions by a mixture of ground burnt patties and red soil. *Advances in Environmental Research*, 3(3), pp. 231 - 251.
- Sadegh, H., Yari, M., Ghoshekandi, R. S., Ebrahimiasl, S., Maazinejad, B., Jalili, M. & Chahardori, M. (2014). Dioxins: a review of its environmental risk. *Pyrex Journal of Research in Environmental Studies*, 1(1), pp. 1 - 7.
- Saltali, K., Sari, A. & Aydin, M. (2007). Removal of ammonium ion from aqueous solution by natural Turkish (Yildizeli) zeolite for environmental quality. *Journal of Hazardous Materials*, 141(1), pp. 258 - 263.

- Sarioglu, M. (2005). Removal of ammonium from municipal wastewater using natural Turkish (Dogantepe) zeolite. *Separation and Purification Technology*, 41(1), pp. 1 - 11.
- Sartaj, M. & Fernandes, L. (2005). Adsorption of boron from landfill leachate by peat and the effect of environmental factors. *J. Environ. Eng. Sci.*, (4), pp. 19 – 28.
- Sayed, Y. E. & Bandosz, T. J. (2004). Adsorption of valeric acid from aqueous solution onto activated carbons: role of surface basic sites. *Journal of Colloid and Interface Science*, 273(1), pp. 64 - 72.
- Şchiopu, A. M. & Ghinea, C. (2013). Municipal solid waste management and treatment of effluents resulting from their landfilling. *Environmental Engineering and Management Journal*, 12(8), pp. 1699 - 1719.
- Seco, A., Marzal, P., Gabaldon, C. & Ferrer, J. (1999). Study of adsorption of Cd and Zn onto an activated carbon: influence of pH, cation concentration, and adsorbent concentration. *Sep. Sci. Technol.*, 34, pp. 1577 – 1593.
- Segui, P., Aubert, J. E., Husson, B. & Measson, M. (2012). Characterization of wastepaper sludge ash for its valorization as a component of hydraulic binders. *Applied Clay Science*, 57, pp. 79 – 85.
- Shahmohamadi-Kalalagh, S., Babazadeh, H., Nazemi, A. H. & Manshouri, M. (2011). Isotherm and kinetic studies on adsorption of Pb, Zn and Cu by kaolinite. *Caspian Journal of Environmental Sciences*, 9(2), pp 243 – 255.
- Shehzad, A., Bashir, M. J. K., Sethupathi, S. & Lim, J. W. (2015). An overview of heavily polluted leachate treatment using food waste as an alternative and renewable source of activated carbon. *Process Safety and Environmental Protection*, 98, pp. 309 - 318.
- Shukla, P. R., Wang, S., Ang, H. M. & Tade, M. O. (2009). Synthesis, characterisation, and adsorption evaluation of carbon-natural-zeolite composites. *Advanced Powder Technology*, 20(3), pp. 245 - 250.

- Singh, S., Srivastava, V. C. & Mall, I. D. (2009). Fixed-bed study for adsorptive removal of furfural by activated carbon. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 332(1), pp. 50 - 56.
- Siswoyo, E. & Tanaka, S. (2013). Development of eco-adsorbent based on solid waste of paper industry to adsorb cadmium ion in water. *Journal of Clean Energy Technologies*, 1(3), pp. 198 - 201.
- Sivakumar, D. (2013). Adsorption study on municipal solid waste leachate using Moringa oleifera seed. *Journal of Environmental Science and Technology*, 10(1), pp. 113 - 124.
- Slack, R. J., Gronow, J. R. & Voulvoulis, N. (2005). Household hazardous waste in municipal landfills: contaminants in leachate. *Science of the Total Environment*, 337(1-3), pp. 119 - 137.
- Steenhuis, T. S., Rivera, J. C., Hernández, C. J. M., Walter, M. T., Bryant, R. B. & Nektarios, P. (2001). Water repellency in New York State soils. *International Turfgrass Society Research Journal*, 9, pp. 624 - 628.
- Stoeckli, F. & Cleary, D. H. (2001). On the mechanisms of phenol adsorption by carbons. *Russian Chemical Bulletin*, 50(11), pp. 2060 - 2063.
- Suzuki, M. (1990). *Adsorption Engineering*. Kodansha-Tokyo: Elsevier Science.
- Świątek, M. Z. & Malińska, K. (2010). Removal of ammonia by clinoptilolite. *Global NEST Journal*, 12(3), pp. 256 - 261.
- Syafalni, S., Abustan, I., Dahlan, I., Wah, C.K. & Umar, G. (2012). Treatment of dye wastewater using granular activated carbon and zeolite filter. *Modern Applied Science*, 6(2), pp. 37 - 51.
- Tatsi, A. A. & Zouboulis, A. I. (2002). A field investigation of the quantity and quality of leachate from a municipal solid waste landfill in a Mediterranean climate (Thessaloniki, Greece). *Advances in Environmental Research*, 6(3), pp. 207 - 219.

- Temel, F. A. & Kuleyin, A. (2016). Ammonium removal from landfill leachate using natural zeolite: kinetic, equilibrium, and thermodynamic studies. *Desalination and Water Treatment*, 57(50), pp. 23873 - 23892.
- Thomas, H. C. (1944). Heterogeneous Ion Exchange in a Flowing System. *Journal of the American Chemical Society*, 66(10), pp. 1664 - 1666.
- Thomas, J. M. dan Thomas, W. J. (1997). *Principle and Practice of Heterogeneous Catalysis*. Weinheim: VCH.
- Toles, C.A., Marshall, W.E., John, M.M., Wartelle, L.H. & McAloon, A. (2000). Acid-activated carbons from almond shells: physical, chemical and adsorptive properties and estimated cost of production. *Bioresource Technology*, 71(1), pp. 87 - 92.
- Toretta, V., Ferronato, N., Katsoyiannis, I. A., Tolkou, A. K. & Airoidi, M. (2016). Novel and conventional technologies for landfill leachates treatment: a review. *Sustainability*, 9(9), pp. 1 - 39.
- Torres, A. L., Correa, E. M. C., González, C. F., Franco, M. F. A. & Serrano, V. G. (2012). On the use of a natural peat for the removal of Cr (VI) from aqueous solutions. *Journal of Colloid and Interface Science*, 386(1), pp. 325 - 332.
- Tyrell, S. F., Leeds-Harrison, P. B. & Harrison, K. S. (2002). Removal of ammoniacal nitrogen from landfill leachate by rogation into vegetated treatment plant. *Water Research*, 36(1), pp. 291 - 299.
- USACE (U.S. Army Corps of Engineers), (2001). Adsorption Design Guide, Engineering and Design. [Online] Department of the Army DG 1110-1-2. Capaian maklumat pada 18 Disember 2014 dari <http://www.publications.usace.army.mil/USACEPublications/EngineerDesignGuides.aspx>
- Villafranco, E. Z., Quintal, I. D. B., Salazar, S. G., Quintal, M. B., Correa, H. E. S. & Rodríguez, J. M. S. (2014). Adsorption kinetics of matter contained in a leachate

using eggshell and activated carbon. *Journal of Environmental Protection*, 5(7), pp. 608 - 619.

Wajima, T. & Rakovan, J. F. (2013). Removal behavior of phosphate from aqueous solution by calcined paper sludge. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 435, pp. 132 - 138.

Wang, S., Zhu, Z. H., Coomes, A., Haghseresht, F. & Lu, G. Q. (2005). The physical and surface chemistry characteristic of activated carbon and the adsorption of methylene blue from wastewater. *Journal of Colloid and Interface Science*, 284(2), pp. 440 - 446.

Wang, S. & Wu, H. (2006). Environmental-benign utilisation of fly ash as low-cost adsorbents. *Journal of Hazardous Materials*, 136(3), pp. 482 - 501.

Wang, Y. -F., Lin, F. & Pang, W. -Q. (2007). Ammonium exchange in aqueous solution using Chinese natural clinoptilolite and modified zeolite. *Journal of Hazardous Materials*, 142(1-2), pp. 160 - 164.

Wang, Z. P., Zhang, Z., Lin, Y. J., Deng, N. S., Tao, T. & Zhuo, K. (2002). Landfill leachate treatment by a coagulation-photooxidation process. *Journal of Hazardous Materials*, 95(1-2), pp. 153 - 159.

Weatherley, L. R. & Miladinovich, N. D. (2004). Comparison of the ion exchange uptake of ammonium ion onto New Zealand clinoptilolite and mordenite. *Water Research*, 38(20), pp. 4305 - 4312.

Weber, W. J. & Morris, J. C. (1963). Kinetics of adsorption on carbon from solution. *Journal of the Sanitary Engineering Division, American Society of Civil Engineering*, 89(1), pp. 31 - 60.

Widiastuti, N., Wu, H., Ang, M. & Zhang, D. (2011). Removal of ammonium from greywater using natural zeolite. *Desalination*, 277(1-3), pp. 15 - 23.

- Wiszniewski, J., Miksch, K., Robert, D., Surmacz-Górska, J. & Weber, J. V. (2006). Landfill leachate treatment methods: a review. *Environmental Chemistry Letters*, 4(1), pp. 51 - 61.
- Worch, E. (2012). *Adsorption Technology in Water Treatment: Fundamentals, Processes, and Modeling*. Berlin/Boston: Walter de Gruyter GmbH and Co. KG.
- Wu, P., Tang, Y., Wang, W., Zhu, N., Li, P., Wu, J., Dang, Z. & Wang, X. (2011). Effect of dissolved organic matter from Guangzhou landfill leachate on sorption of phenanthrene by montmorillonite. *Journal of Colloid and Interface Science*, 361(2), pp. 618 - 627.
- Xu, Z. Y., Zeng, G. M., Yang, Z. H., Xiao, Y., Cao, M., Sun, H. S., Ji, L. L. & Chen, Y. (2010). Biological treatment of landfill leachate with the integration of partial nitrification, anaerobic ammonium oxidation and heterotrophic denitrification. *Bioresource Technology*, 101(1), pp. 79 - 86.
- Yang, X., Yi, H., Tang, X., Zhao, S., Yang, Z., Ma, Y., Feng, T. & Cui, X. (2018). Behaviors and kinetics of toluene adsorption-desorption on activated carbons with varying pore structure. *Journal of Environmental Sciences*, 67, pp. 104 - 114.
- Yao, P. (2013). Perspective on technology for landfill leachate treatment. *Arabian Journal of Chemistry*, 10(2), pp. 2567 - 2574.
- Yin, C. Y., Syed Abdul Kadir, S. A., Abdul-Malik, A. S., Syed-Arifin, S. N., Mahazer, S. Z. & Zamzuri, Z. (2007). Characterization of ash derived from combustion of paper mill waste sludge: comparison with municipal solid waste incinerator ash. *ScienceAsia*, 33(4), pp. 473 - 477.
- Yoon, Y. H. & Nelson, J. H. (1984). Application of gas adsorption kinetics: Part 1, A theoretical model for respirator cartridge service time. *American Industrial Hygiene Association Journal*, 45(8), pp. 509 - 516.
- Yuan, Y. & Lee, T. R. (2013). *Contact Angle and Wetting Properties*. Springer Series in Surface Sciences.

- Zaini, M. S., Ismail, N., Lim, C. K., Neoh, C. H., Lam, C. Y., Aris, A., Majid, Z. A., Manan, F. A. & Ibrahim, Z. (2015). Optimisation of biostructure for the adsorption of petrochemical wastewater using statistical approach. *Clean Technologies and Environmental Policy*, 17(1), pp. 249 - 256.
- Zakaria, N. F., Majid, Z. A., Ramli, Z., Jaafar, J., Aris, A., Talib, J. & Ali, R. (2016). Adsorbent from waste and natural deposits for paraquat removal in water. *Malaysian Journal of Analytical Sciences*, 20(3), pp. 469 - 476.
- Zendelska, A., Golomeova, M., Blazev, K., Krstev, B., Golomeov, B. & Krstev, A. (2014). Kinetic studies of zinc ions removal from aqueous solution by adsorption on natural zeolite. *International Journal of Science Environment*, 3(4), pp. 1303 - 1318.
- Zin, N. S. M., Aziz, H. A., Adlan, M. N., Ariffin, A., Yusoff, M. S. & Dahlan, I. (2014). Treatability study of partially stabilized leachate by composite coagulant (prehydrolyzed iron and tapioca flour). *International Journal of Scientific Research in Knowledge*, 2(7), pp. 313 - 319.
- Zouboulis, A. I., Chai, X. -L. & Katsoyiannis, I. A. (2004). The application of biofloculant for the removal of humic acids from stabilized landfill leachates. *Journal of Environmental Management*, 70(1), pp. 35 - 41.

